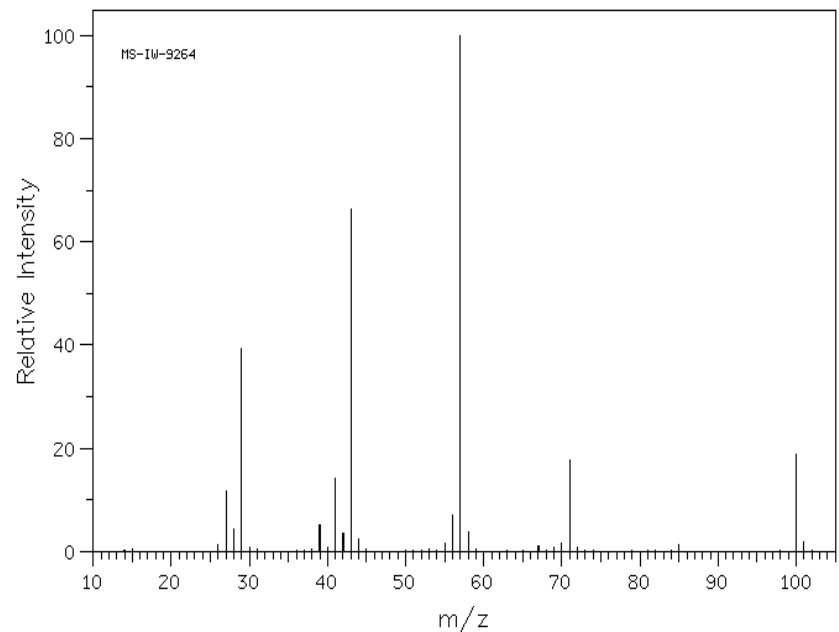
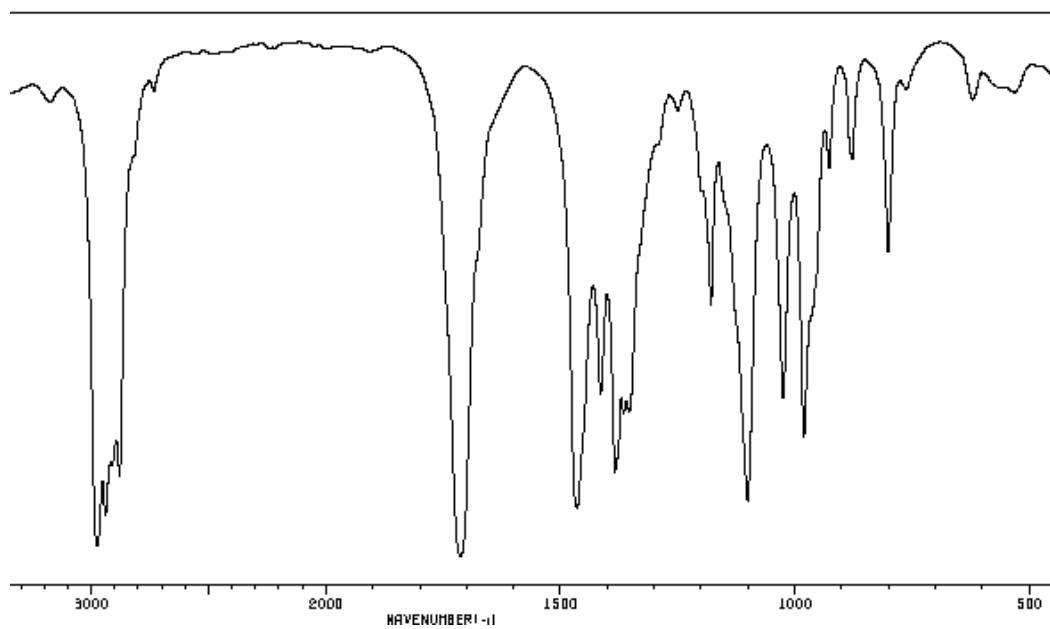
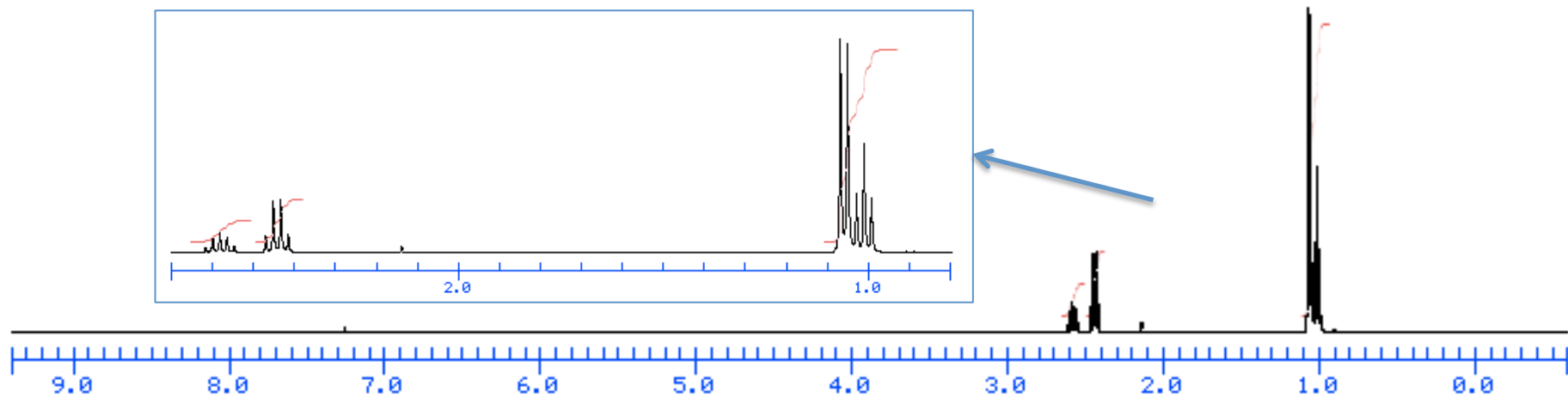
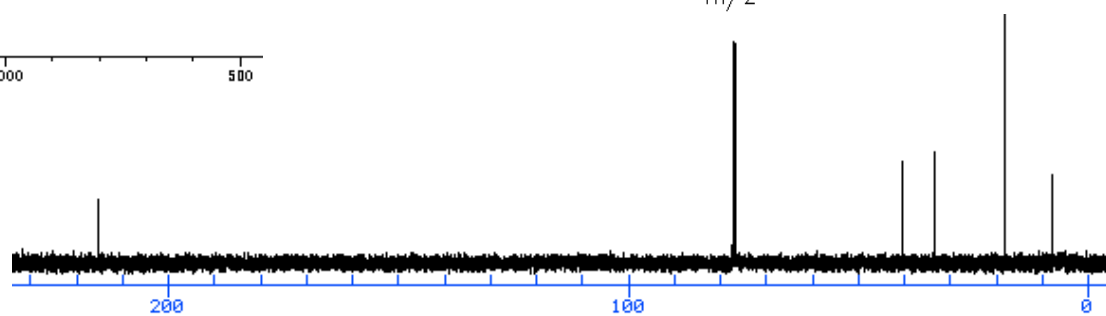


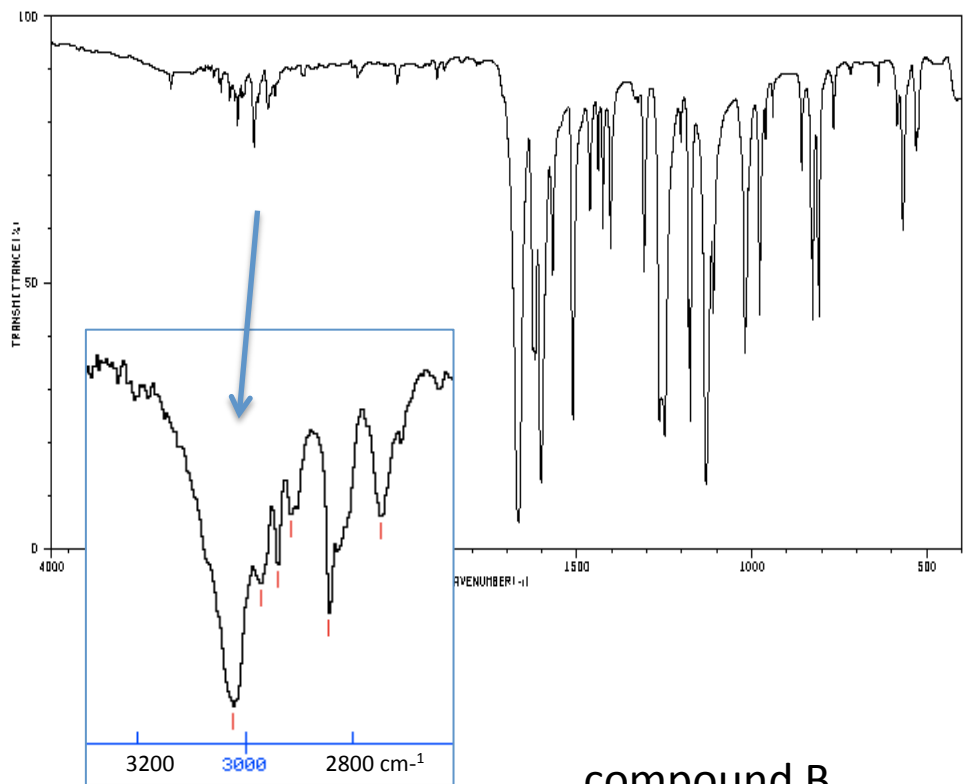
MW to molecular formula

- If even; no halides
 - If hydrocarbon – $\#C = (\text{mol wt.} - 2) / 14$ (round UP!)
 - Calculate IHD from $\#C$ and $\#H$
 - Alternate formulas with O: Add in O, take away CH_4 , increase IHD by 1
 - Alt. formulas with 2 N: Add in N_2 , take away C_2H_4 , increase IHD by 1
 - Alt. formulas with S: Add in S, take away C_2H_8 , increase IHD by 2
 - If IHD becomes 7 or more, consider alt. with one fewer C, 12 more H
- If one halide
 - Detect by $M+2$ peaks (Cl, Br) or $M-19$ pk (F; may be small) or $M-127$ (I)
 - Subtract 18, 34, 78 or 126 from M to find wt. if halide replaced by H
- If odd then odd number of nitrogens (1, 3 or 5)
 - Subtract 15, 45 or 75 from M
 - determine formula as above
 - add in NH , N_3H_3 or N_5H_5

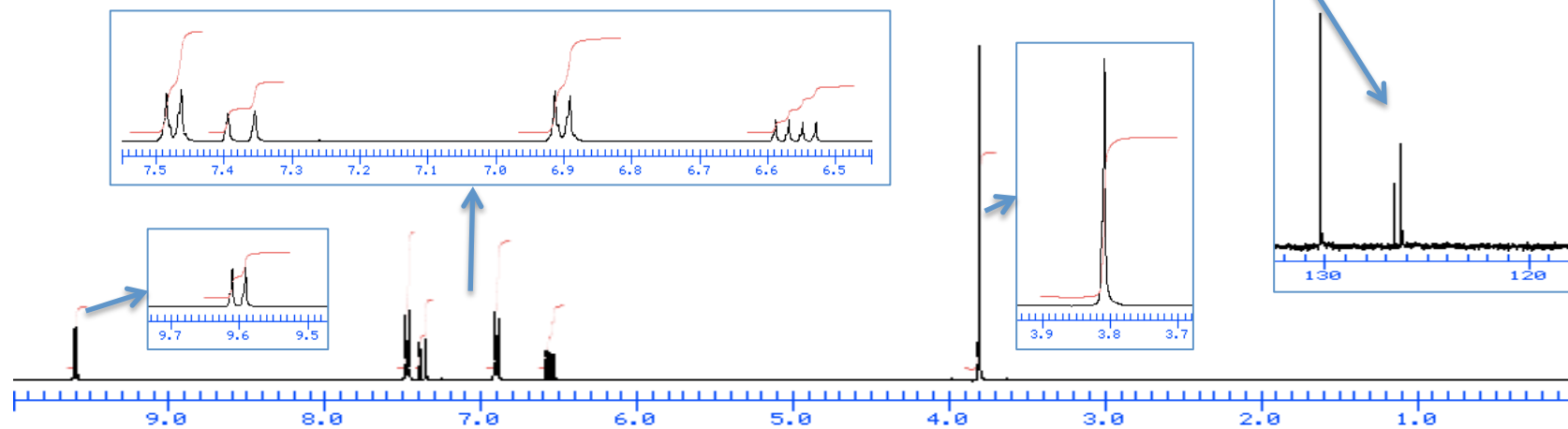
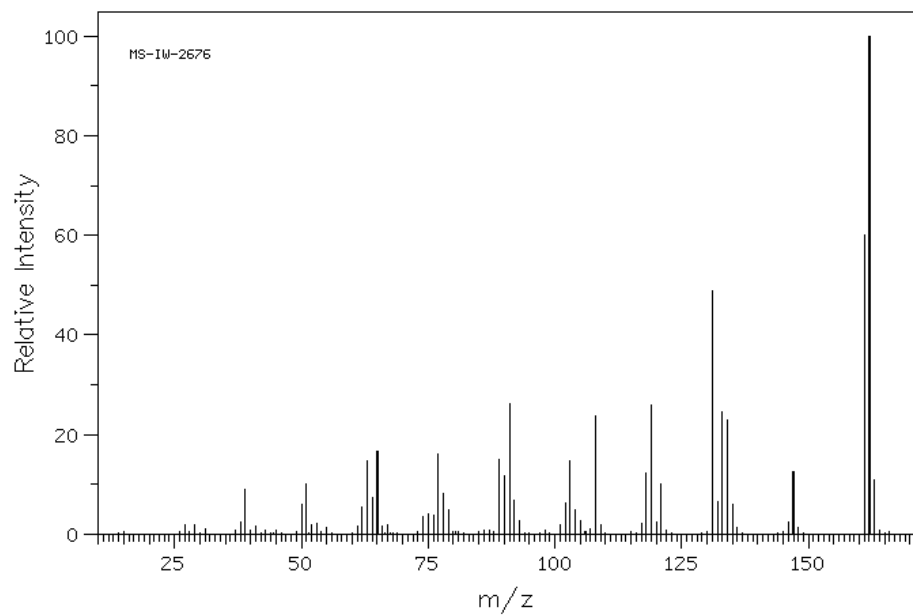


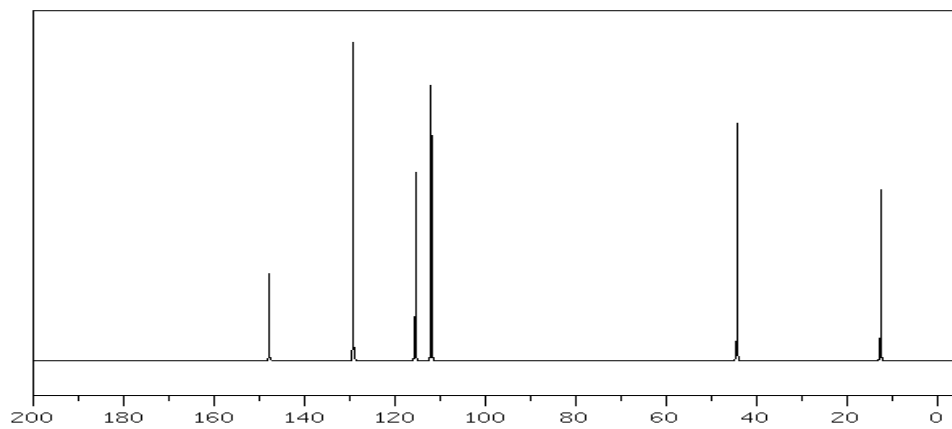
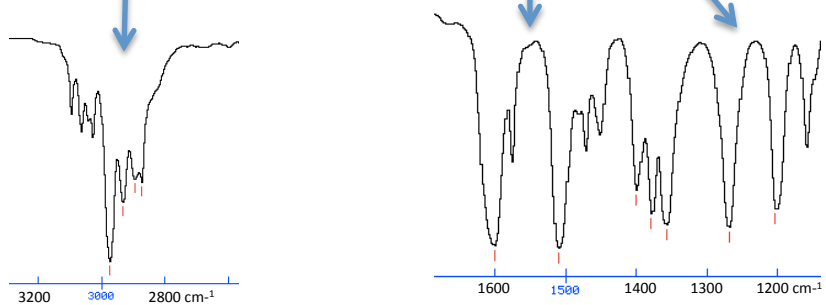
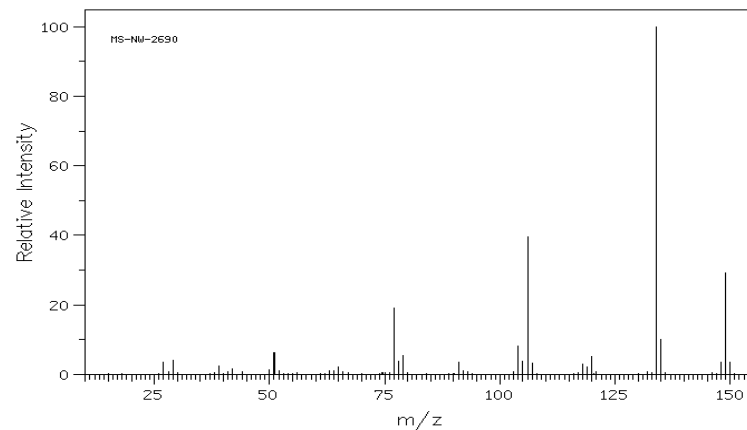
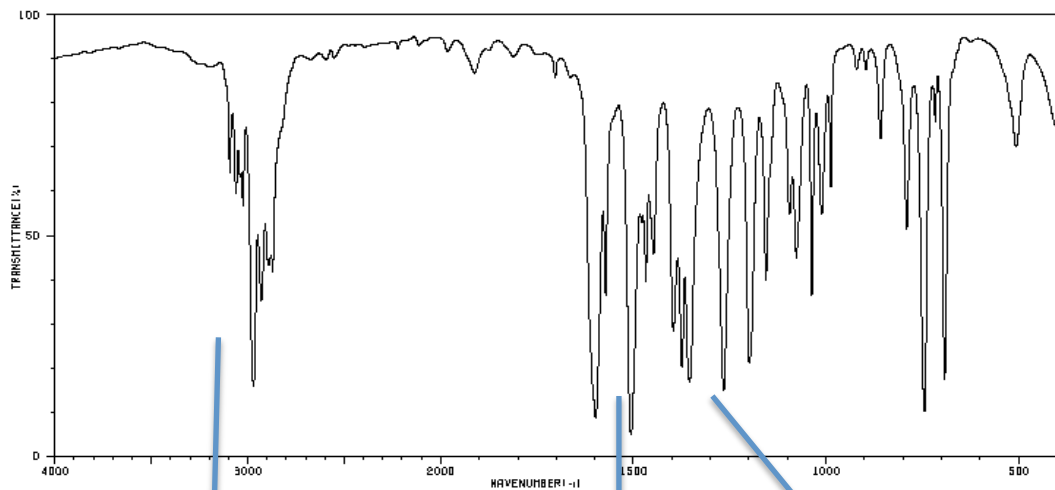
compound A





compound B





unknown C

